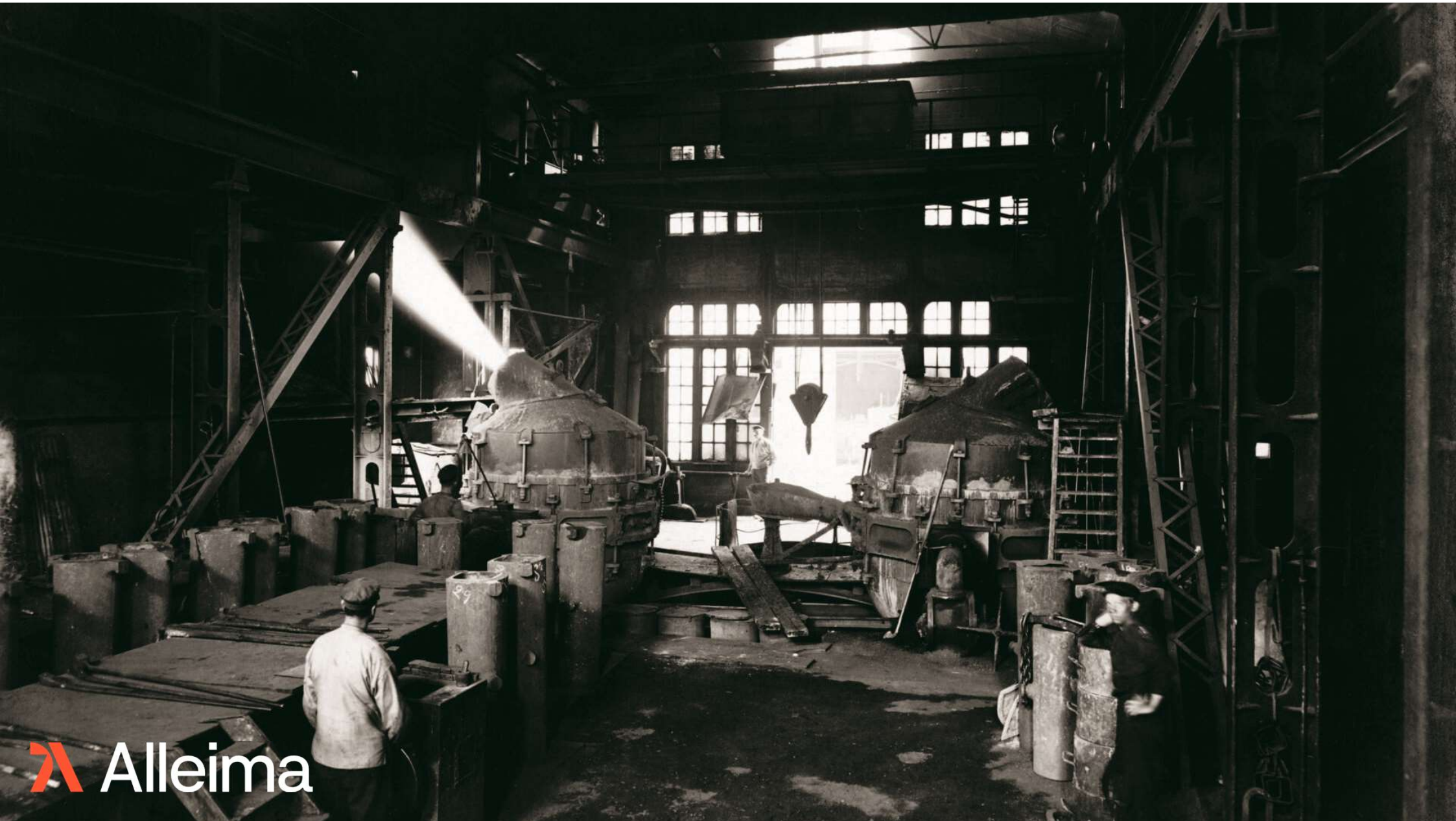




# Duplex Stainless Steel in Heat Exchangers for Alkaline Electrolyzers

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 Alleima



# Advancing industries through materials technology



## Industrial

- Solid round bar and hollow bar
- High pressure tubing
- Composite tubing
- Wear resistance strip



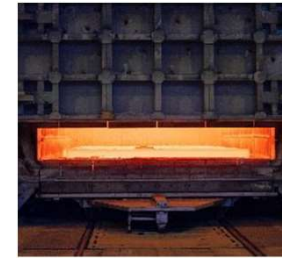
## Chemical and petrochemical

- Fertilizer tubing
- Hydraulic and instrumental tubing
- Heat exchanger tubing



## Oil and gas

- Umbilical tubing
- Control lines
- Oil Country Tubular



## Industrial heating

- Metallic heating elements
- Ceramic heating elements
- Radiant Kanthal® APM / Kanthal APMT® tubes
- Diffusion cassettes



## Consumer

- Compressor valve steel
- Stainless knife steel
- Razor blade steel
- Appliance wire



## Power generation

- Steam generator tubes
- Cladding tubing
- Nuclear tubes and pipes
- Strip steel spacers



## Mining and construction

- Rock drill steel



## Transportation

- Titanium and stainless-steel tubes
- Gasoline Direct Injection (GDI) tubes
- Compressor valve steel
- Shock absorber steel



## Medical

- Medical precision wire
- Medical tubing
- Medical strip



## Hydrogen and renewable energy

- Coated strip steel for fuel cells
- High pressure tubing for hydrogen applications
- High nickel alloy tubing for concentrated solar power applications

# Agenda



# Agenda





The image is a 3D cutaway diagram of an alkaline electrolyzer system. It shows a large rectangular electrolyzer cell with internal components like electrodes and a central stack. To the left, there are electrical control units. To the right, there are two large vertical cylindrical storage tanks connected by pipes. A large semi-transparent blue circle is overlaid on the center of the diagram, containing the title text. A dashed orange line curves around the top left of the blue circle. A solid teal circle is located at the bottom right of the blue circle.

## ALKALINE ELECTROLYZER INTRODUCTION

Image used by permission from Nel Hydrogen.

# Alkaline electrolyzers – basic operations

- Electrolyte is commonly 25-30% KOH
- Produces hydrogen and oxygen gas by electrolysis of water in an alkaline environment
  - Cathode reaction:  $2 \text{H}_2\text{O}(\text{l}) + 2\text{e}^- \rightarrow \text{H}_2(\text{g}) + 2 \text{OH}^-(\text{aq})$
  - Anode reaction:  $2 \text{OH}^-(\text{aq}) \rightarrow \frac{1}{2}\text{O}_2(\text{g}) + \text{H}_2\text{O}(\text{l}) + 2\text{e}^-$
  - **Overall reaction:**  $\text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2}\text{O}_2(\text{g})$
- Operating pressure up to 30 bar
- Operating temperature 80-90°C



# Alkaline electrolyzers – design considerations

- Components in the electrolyzer system
  - The cell stack.
  - Piping for electrolyte and electrolyte + gases
  - Gas separators
  - Tanks for electrolyte and water replenishment
  - Heat exchangers for cooling and balance of plant (BOP)
    - Depending on location, freshwater or seawater cooling can be used

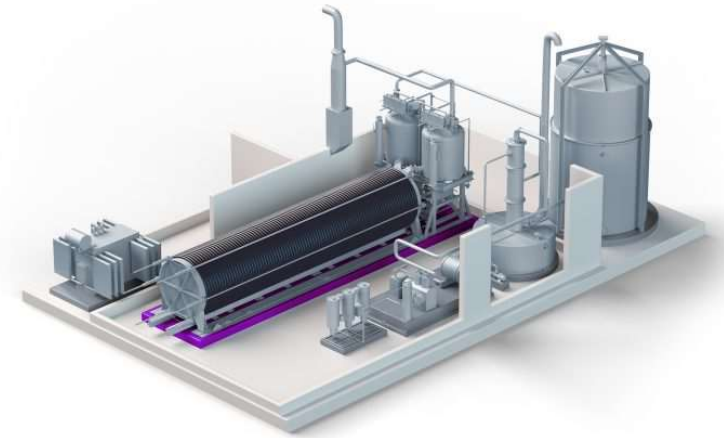


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# Alkaline electrolyzers – design considerations

- Water is consumed in operation and replenished
  - Any added impurities will build up in the system
    - Water needs to be very clean
  - Corrosion products will build up in the system
    - Can cause problems with catalysts
    - Can cause blockage in small-diameter tubes and pipes.
    - **Materials must be corrosion-resistant to the operating environment**



# Agenda

## **Stainless steels in caustic solutions**

General information and chemical composition

Corrosion resistance of austenitic and duplex grades

Conversion of corrosion data for NaOH to KOH

# General information

- Solutions of sodium or potassium hydroxide (NaOH and KOH) are not corrosive to stainless steel as long as
  - Concentration is not too high
  - Temperature is not too high
- Corrosion mechanisms are
  - General corrosion.
  - Stress corrosion cracking
- 304L austenitic stainless steel has been used successfully for many caustic applications in the pulp and paper industry
  - Reasonable long service life and low replacement cost when using a welded pipe



# Stainless steel chemical composition

Typical chemical composition (nominal), %						
Austenitic grades						
Alloy	UNS	C	Cr	Ni	Mo	N
304L	S30403	<0.03	18.5	10	-	
316L	S31603	<0.03	17	11.5	2.1	
904L	S08904	<0.03	20	25	4.5	
Duplex grades						
2304	S32304	<0.03	22.5	4.5	0.3	0.10
2205	S32205	<0.03	22	5	3.2	0.18
2507	S32750	<0.03	25	7	4	0.30
2906	S32906	<0.03	29	7	2.3	0.35



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# Corrosion resistance of austenitic and duplex grades

– General corrosion data for selected stainless steel grades in NaOH

Conc. %	10	10	10	20	20	25	25	30	30	30	40	40
Temp. °C	20	90	103=BP	20	90	20	112=BP	20	100	116=BP	80	90
Carbon steel	0			0		0		0				
13 Cr	0	0	1	0	1	0	2	0	1	2	1	1
304L (1.4306)	0	0	0	0	0	0	0	0	0	1s	0	0
316L (1.4435)	0	0	0	0	0	0	0	0	0	0s	0	0
904L	0	0	0	0	0	0	0	0	0	0s	0	0
Duplex 2304	0			0				0	0	OND	0	0
Duplex 2205	0			0				0	0	OND		
Duplex 2507	0			0				0	0	OND		

0 = <0.1mm/y

1 = 0.1 – 1 mm/y

S = Stress corrosion cracking

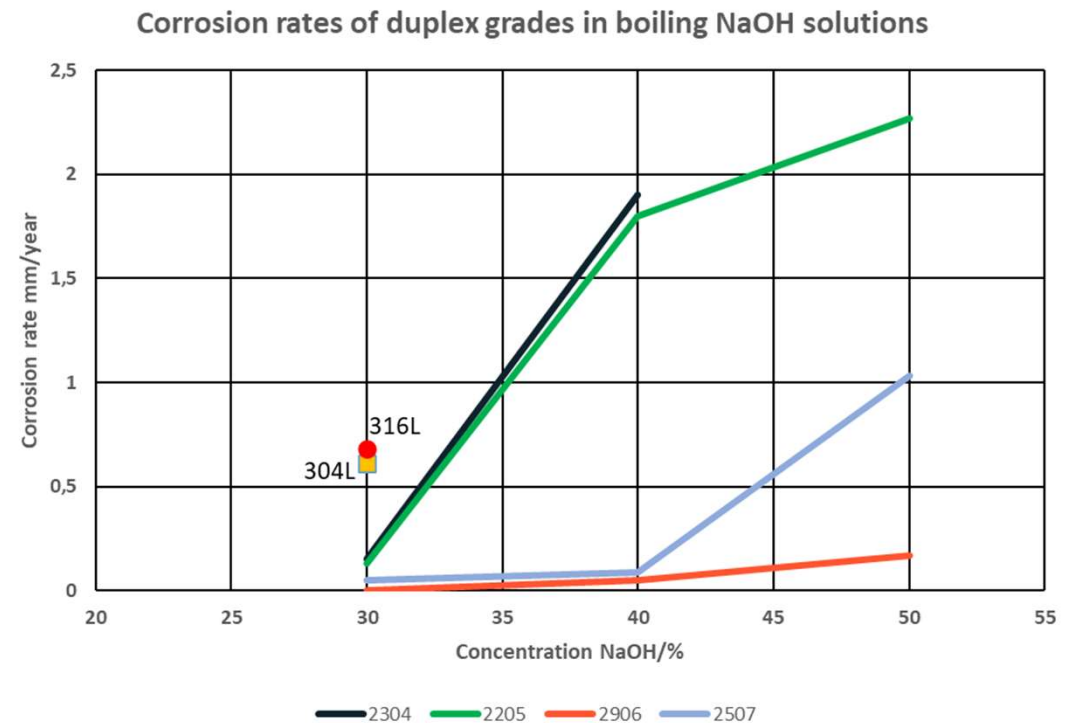
ND = Not tested

BP = Atmospheric Boiling point

Source: Corrosion handbook stainless steels

# Duplex grades in strong caustic solutions

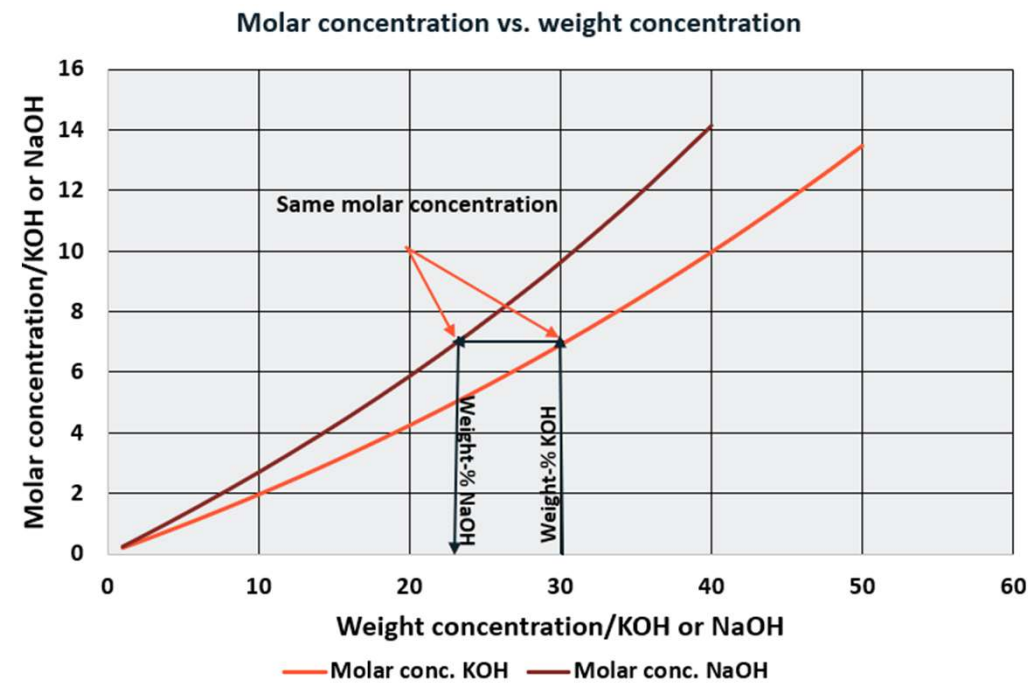
- In-house testing of duplex grades suitable for caustic service
- Duplex 2304 and 2205 show very similar performance
- Duplex 2906 outperforms the other grades.
  - 0.05 mm/y in boiling 40% NaOH
- Results confirm that higher chromium concentration gives better corrosion resistance.



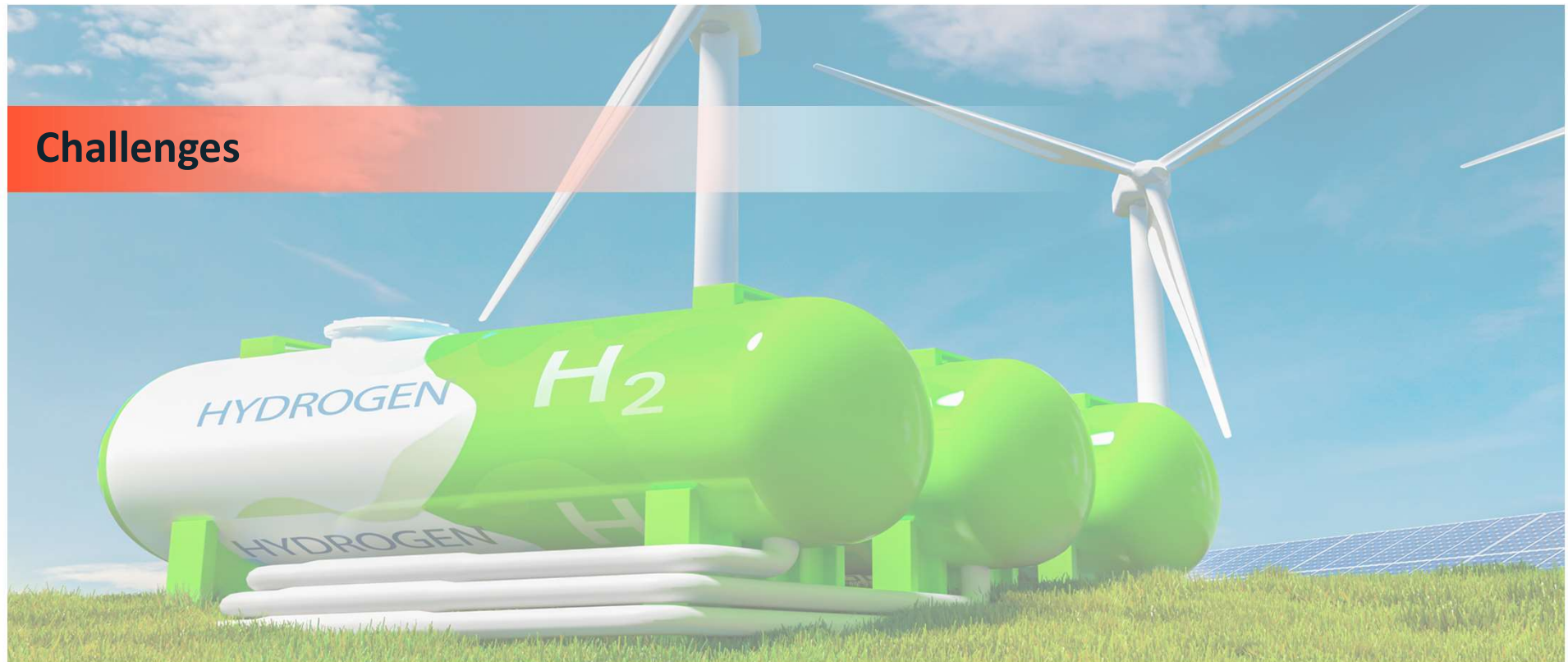


# Conversion of corrosion data for NaOH to KOH

- There is very little published data for corrosion in KOH.
  - Data for NaOH can be used instead.
  - $\text{Na}^+$  and  $\text{K}^+$  do not play an active part in the corrosion process.
- The corrosive part of a caustic solution is the  $\text{OH}^-$  ion.
  - The molar concentration of  $\text{OH}^-$  governs the corrosivity of the solution.
- Potassium is heavier than sodium.
  - 20 weight-% KOH has a lower OH concentration than 20 weight-% NaOH.
  - 20 weight-% NaOH more corrosive than 20 weight-% KOH.
  - 30 weight-% KOH  $\approx$  23 weight-% NaOH.

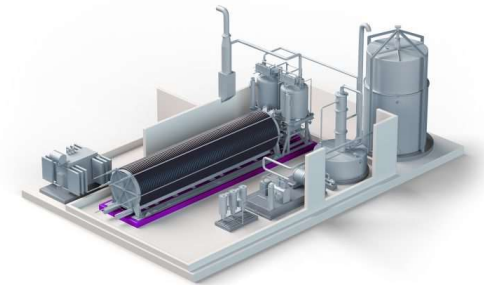


# Agenda



**Challenges**

# Design criteria



- For some alkaline electrolyzer plants, the design criteria are very tough.
  - 40% KOH ( $\approx$  32% NaOH).
    - Up-concentration when water is consumed.
  - $T = 120^{\circ}\text{C}$ .
  - $P = 40$  bar.
- Design life >30 years without any replacement of components...
  - Corrosion rates >0.05 mm/y adds up to >1.5 mm/y.
  - No corrosion products allowed.
    - Can cause fouling of piping system, tanks, and cell stack.

# Parts after electrolyzer stack and for gas separation are exposed to high pressures of

## Hydrogen

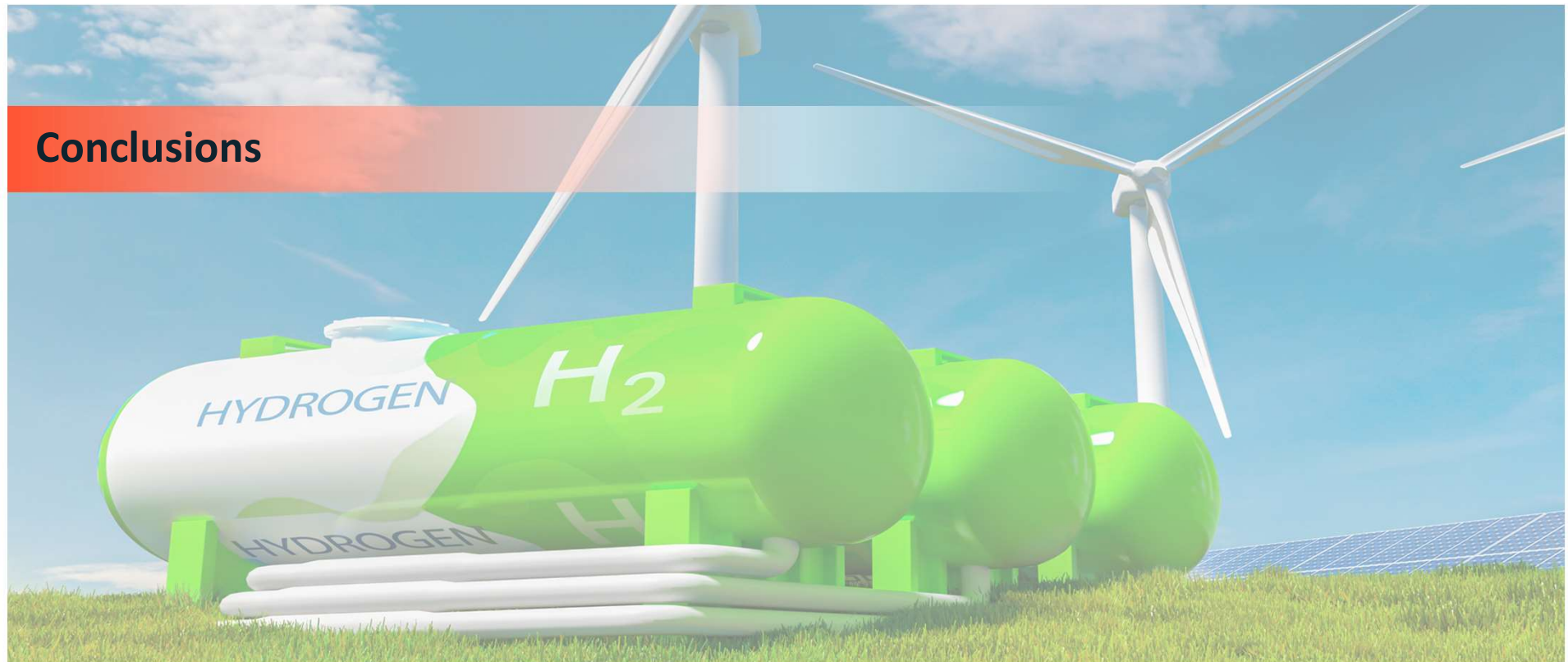
- Cause loss the ductility and embrittlement of
  - Austenitic stainless steels with low nickel content
  - Duplex stainless steels because of the ferrite content
- **Higher alloyed austenitic grades should be considered**

## Oxygen

- Increase the risk for SCC of stainless steel grades
- More tests and investigations are necessary to find the best suitable stainless steel grades



# Agenda



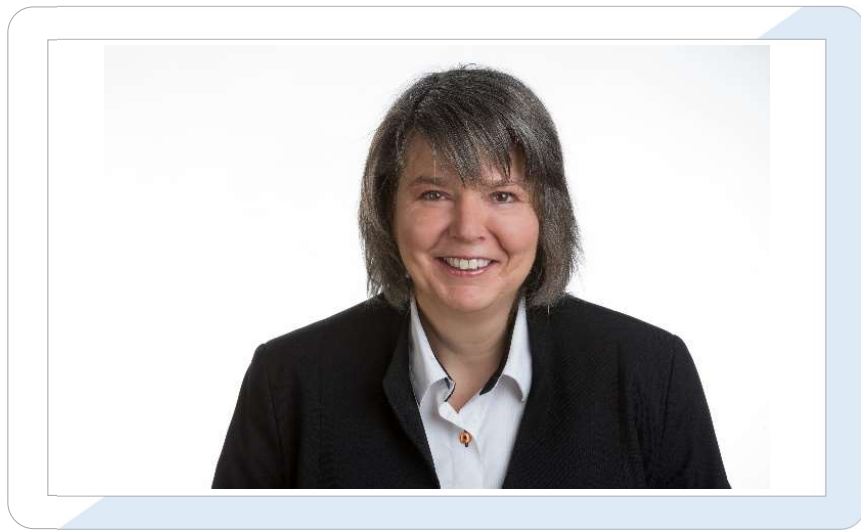
# Conclusions

- Alkaline electrolyzers are well-established in the hydrogen market
  - 300-series stainless steel has served well
- New developments and demands on performance and plant service life make duplex grades highly attractive
- Resistance towards stress corrosion cracking at high oxygen pressures must be investigated
- Suitability of duplex grades in high-pressure hydrogen service is questionable

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